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Stanley, George G. Publisher Brooks/Cole
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in Figure 9.3. Aluminum is a solid, and
bromine is a liquid. The brownish-red
cloud in the photograph is excess
bromine. The reaction's product, which
is solid particles of aluminum oxide
(Al₂O₃), settles on the bottom of the
beaker. Reactant 1 + Reactant 2 →
Product 1 aluminum(s) + bromine(l) →
aluminum bromide(s)

Chapter 9: Chemical Reactions

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Structure of the Atom 2. Quiz: Element
Symbols first 20 elements plus Br and I

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transition metals: Sb, As, Bi, Cd, Cr, Co, Cu, Au, Fe, Pb, Mn, Hg, Ni, Pt, Ag, Sn, Ti, Zn 3. Element symbol Rules for writing the element symbols: First letter capitalize Second lower case Hints ...

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3H 4; 3 mol CO 2:1 mol C 3H 4; 2 mol H 2O:1 mol C 3H 4; 3 mol CO 2:4 mol O 2;

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Answers

2 mol H₂O:4 mol O₂; 2 mol H₂O:3 mol CO₂; or their reciprocals
b. Suppose that x moles of water form in the above reaction. The other three mole quantities (not in order) are $2x$, $1.5x$, and $0.5x$. Match these quantities to their respective components in the equation

...

mc06se cFMsr i-vi

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Chemical Reviews

OBJ: 9.3.1 Interpret the prefixes in the names of molecular compounds in terms of their chemical formulas. | 9.3.2 Apply the rules for naming and writing formulas for binary molecular compounds. STA: Ch.2.a 22.ANS: C PTS: 1 DIF: L1 REF: p. 269 OBJ: 9.3.2 Apply the rules for naming and writing formulas for binary molecular compounds. STA: Ch.2

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Chapter 9 Practice Test - Naming and Writing Chemical Formulas

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